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The use of a novel microwave precipitation technique for the synthesis of bioactive nanoparticles

John Noon, Tom Mills, Aditya Nayak, Ian Norton

Department of Chemical Engineering, The University of Birmingham, Edgbaston,
Birmingham, B15 2TT



University of
Nottingham
UK | CHINA | MALAYSIA



Loughborough
University

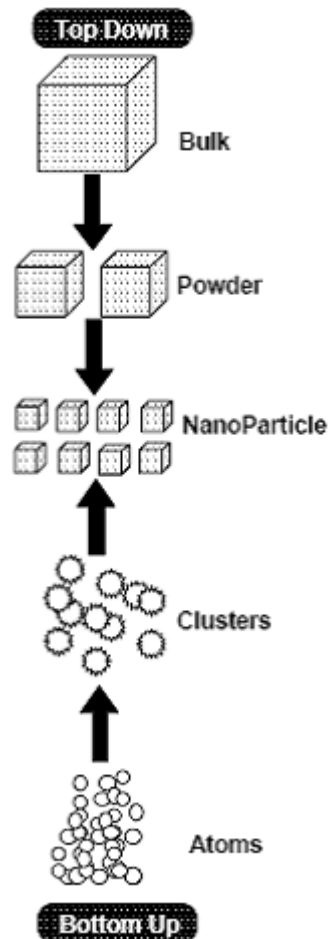
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Bioactive compounds

- These are the extra-nutritional components in food which are capable of providing numerous health benefits upon consumption.
- A wide variety of Bioactive Compounds (BC's) can be found in nature.
- However, unfortunately many BC's suffer limited oral bioavailability which is often due to their poor aqueous solubility.

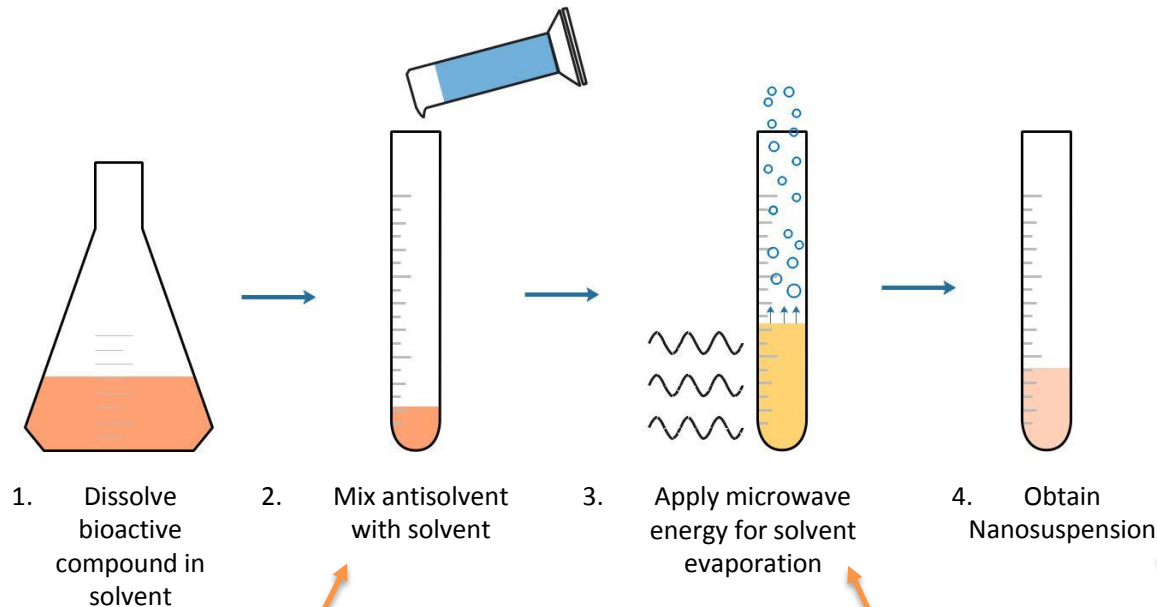
Nanonisation Techniques



- Top Down Techniques- E.g High Pressure homogenisation and milling.
- These techniques are generally well established and typically capable of processing large throughputs with a high energy demand.
- Bottom Up Techniques- E.g. Antisolvent precipitation, and spontaneous emulsification.
- These techniques are generally less energy intensive and give greater control over particle properties.

Figure 1: Top Down Vs Bottom Up

Microwave Antisolvent Precipitation (MAP) Technique



Huge reduction in solvent quality and 'first wave' of supersaturation. However the reaction is incomplete.

Rapid solvent removal through microwave heating further reduces solvent quality and generates a 'second wave' of high supersaturation.

Nanosuspension has greatly enhanced physical stability due to solvent removal.

In this study:

Bioactive Compound- **Curcumin**

Solvent- **Ethanol**

Antisolvent- **Water**

Effect of Processing and Formulation Parameters on Particle Size



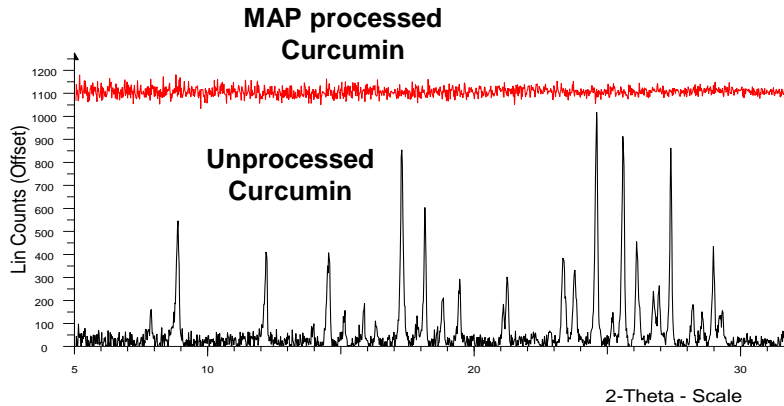
Pressure (kPa)	Average Size (nm)	PDI	Concentration (mg/ml)	Average Size (nm)	PDI
0	191 ±6	0.15	0.5	ND*	ND*
-90	179 ±1	0.06	1	254 ±84	0.40
-93	169 ±6	0.07	2	165 ±2	0.18
-96	148 ±10	0.07	4	148 ±10	0.09
Power-Time Split	Average Size (nm)	PDI	SAS Ratio	Average Size (nm)	PDI
20W, 10mins	159 ±13	0.09	1:7.5	235 ±30	0.07
50W, 4mins	148 ±10	0.07	1:10	169 ±2	0.14
100W, 2mins	135 ±3	0.13	1:20	152 ±4	0.06
200W, 1 mins	148 ±2	0.15	1:30	148 ±10	0.07

*ND- Not Determined

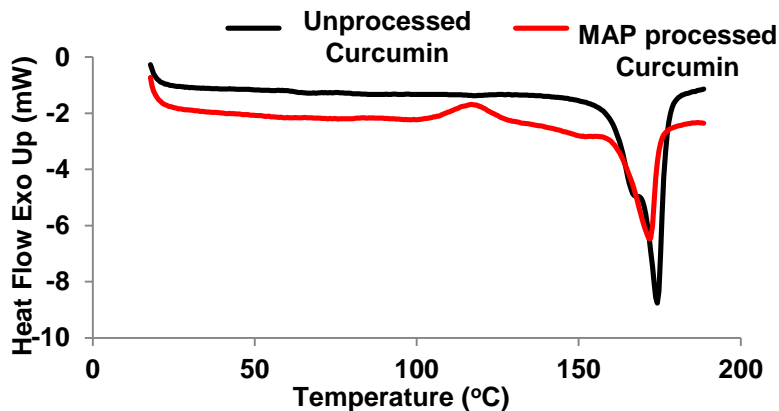
Solid-State Characterisation



- X-ray Powder Diffraction (XRD) and Differential Scanning Calorimetry (DSC) was performed to assess changes in the degree of crystallinity of curcumin post-processing.

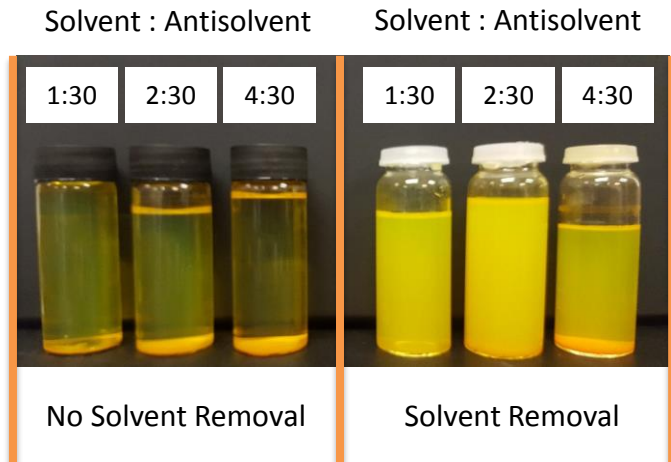
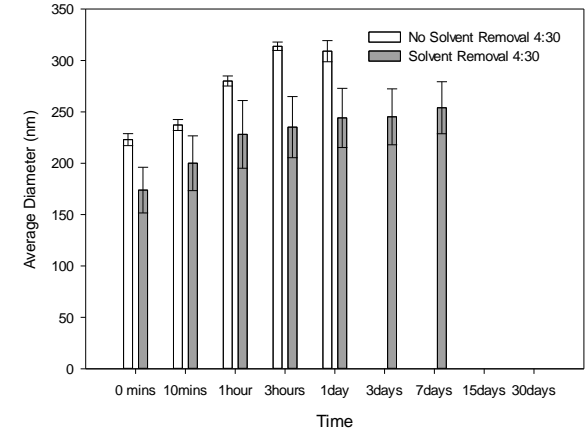
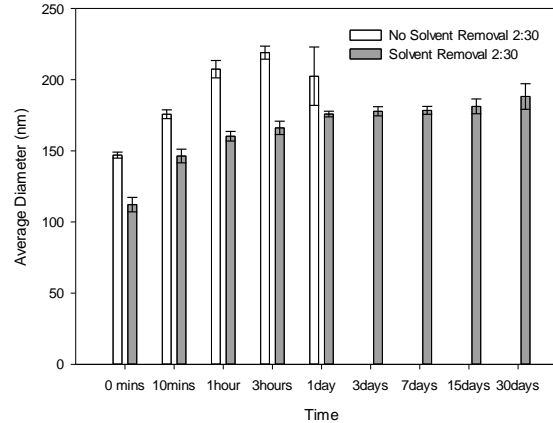
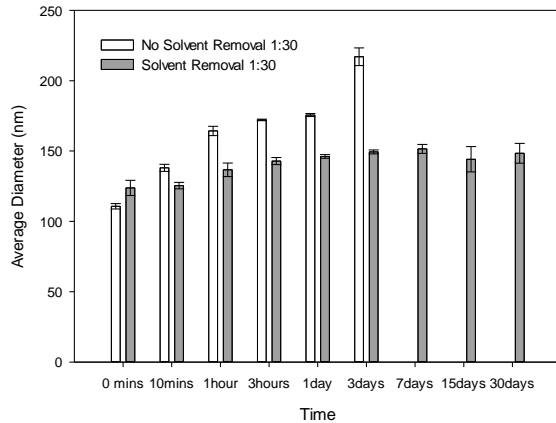


- XRD studies shown the disappearance of the sharp, characteristic crystalline peaks of curcumin once the MAP technique was performed.
- Both the XRD and DSC studies are in agreement with each other and show that curcumin in transformed from a highly crystalline, to a more amorphous morphology post-processing.



- DSC studies showed a significant reduction in the enthalpy of melting of curcumin once the MAP technique was used.

Impact of Solvent Removal on Nanosuspension Stability



Visual appearance of nanosuspensions after 15days

- At all SAS ratio's solvent removal through the MAP technique dramatically improved the physical stability of nanosuspensions from durations of ~1day to in excess of 1 week.
- MAP is a rapid technique which can be completed in a matter of minutes – but, nanosuspension stability can be dramatically improved through use of this process.



New Hypothesis

- The driving force for precipitation comes, almost entirely from the solvent-antisolvent mixing step.
- Solvent removal causes a negligible amount of precipitation in comparison.
- However, solvent removal is very important for the long term stability of curcumin nanosuspensions and can be quickly achieved using the MAP technique.

Conclusions



- Curcumin has been successfully nanosized from an average particle diameter of $\sim 20\mu\text{m}$ to $\sim 0.15\mu\text{m}$.
- Nanosuspensions were produced which exhibited relatively low polydispersity ($\text{PDI} < 0.1$) and significantly improved stability over samples produced in which there was no solvent removal.
- The MAP technique is a rapid process which utilises one of the alternate sources of green energy used in both food and pharmaceutical industries. Thus, this technique could have useful application in both these fields.